

# Chemistry Worksheet

## Mole Calculations

Name: \_\_\_\_\_

Period: \_\_\_\_\_ Date: \_\_\_\_\_

- 1) How many moles are in 15 grams of lithium?
- 2) How many grams are in 2.4 moles of sulfur?
- 3) How many moles are in 22 grams of argon?
- 4) How many grams are in 88.1 moles of magnesium?
- 5) How many moles are in 2.3 grams of phosphorus?
- 6) How many grams are in 11.9 moles of chromium?
- 7) How many moles are in 9.8 grams of calcium?
- 8) How many grams are in 238 moles of arsenic?
- 9) How many moles are in 30 grams of lithium?
- 10) How many atoms are in 12.4 moles of tin?
- 11) How many atoms are in 22.7 grams of radon?
- 12) How many atoms are in 88.19 moles of tungsten?
- 13) How many grams are in  $2.23 \times 10^{24}$  atoms of iron?
- 14) How many moles are in  $1.19 \times 10^{23}$  atoms of sodium?
- 15) What is the mass of  $19.82 \times 10^{23}$  atoms of gold?
- 16) How many moles is  $23.8 \times 10^{23}$  atoms of silver?

4.3	$5.309 \times 10^{25}$	$6.486 \times 10^2$	2.2	$5.5 \times 10^{-1}$	$6.19 \times 10^2$	$2.4 \times 10^{-1}$	$7.46 \times 10^{24}$
3.95	$2.07 \times 10^2$	$7.7 \times 10^1$	$7.4 \times 10^{-2}$	$1.78 \times 10^4$	$2.14 \times 10^3$	$6.16 \times 10^{22}$	$1.98 \times 10^{-1}$